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| Course | CHEM 2401-001 / Quantitative Chemical Analysis (“QCA”) |
| Professor | Dr. Dushanthi Dissanayake |
| Term | Fall 2019 |
| Meetings | Mondays & Wednesdays / 12:00 – 3:45 pm |

Professor’s Contact Information

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|---|---|
| Office Phone | 972-883-3992 |
| Office Location | Berkner Hall (BE) BE 2.316 |
| Email Address | dxd122030@utdallas.edu |
| Office Hours | Mondays 10-11 am and Fridays 2-3 pm and by appointment |
| Other Info: | <p style="text-align: center;"><u>CHEM 2401 / Section 001 (Mondays & Wednesdays)</u></p> SLC 2.302 12:00 – 1:15 pm <i>Notes: We read Emails only from</i> BE 3.102 1:15 – 3:45 pm <i>____@utdallas.edu senders; and,</i> BE 2.506 1:15 – 3:45 pm <i>We do NOT read/use WebCT,</i> <i>eLearning, BlackBoard, etc.</i> |
| Rooms & Times & Teaching Assistant | <p>TA: Yafen Tian Email Address: yxt180000@utdallas.edu Office Hour: TBD Office Hour Location: TBD</p> |

General Course Information

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| Pre-Requisites | CHEM 1312 and 1112 (General Chemistry II Lecture and Lab). |
| Course Description | A study of theories, applications, and calculations involved in methods of analysis, and the practice of volumetric, gravimetric, and spectrophotometric methods. |
| Learning Outcomes | <p><i>Objectives:</i> This course emphasizes the theory, applications, calculations, and practice involved in volumetric, gravimetric, and spectrophotometric methods of analysis (in other words: “What Analytical Chemists Do”).</p> <p><i>Expected Learning Outcomes</i> Students should be able to:</p> <ol style="list-style-type: none">1. Solve stoichiometric and other analytical calculations2. Demonstrate their ability to carry out quantitative volumetric, photometric, and potentiometric determinations3. Explain the necessity for and use of error estimates and statistical methods4. Master the use of spreadsheets like Excel5. Operate at a level of good laboratory practice including safety and cleanliness6. Implement a professional-level lab notebook7. Construct professional-level lab reports |
| Required Materials | <ul style="list-style-type: none">• Quantitative Chemical Analysis, 7th, 8th, 9th, or 10th ed.” by Daniel C. Harris http://bcs.whfreeman.com/webpub/chemistry/qca9e/Catalog_Resources/QCA9eStudentExperimentsforwebsite29Mar2015.pdf• One pad lock (combination or keyed) and a big folder/binder for handouts.• “Cold Springs Harbor Research Laboratory Notebook (NB)” This 8.5”x11” NB was chosen since it has <i>carbon-copy pages</i>. |
| Supplemental Material & Info | <ul style="list-style-type: none">• The Office of Undergraduate Education https://oue.utdallas.edu/• Gen Chem TA Office Hours: visit SLC 3.409 for their weekly schedules.• Please go to the Chem. Dept. Mailroom (BE 2.502) for a list of private <u>tutors</u>. |

Schedule & Academic Calendar

| <u>Meeting</u> | <u>Date</u> | <u>Lecture and/or Activity</u> | <u>Lab Exp.#</u> | <u>Due Dates</u> |
|-------------------|-------------------|--|-------------------------|-------------------------|
| 01 M | 8/19 | Welcome to the World of Analytical Chemistry /Schedules/Grading/Lab Safety | | |
| 02 W | 8/21 | Linear Regression / Excel & Word | Drawers | |
| 03 M | 8/26 | Pipette Calibrations / Lab NoteBooks | Calibrate Your Pipettes | |
| 04 W | 8/28 | Buret Calibrations / Lab Reports | Calibrate Your Pipettes | |
| M | 9/02 | Labor Day Holiday | ----- | |
| 05 W | 9/04 | Statistics Lectures | Calibrate Your Buret | |
| 06 M | 9/09 | Experiment #6 / Statistics Lectures | Calibrate Your Buret | |
| 07 W | 9/11 | Acids, Bases, Buffers, Titrations | Exp. 6 | Buret Graph / Stat Quiz |
| 08 M | 9/16 | Acids, Bases, Buffers, Titrations | Exp. 6 | |
| 09 W | 9/18 | Lab Reports / Midterm Problems | ----- | |
| 10 M | 9/23 | Acids, Bases, Buffers, Titrations | Exp. 8 | |
| 11 W | 9/25 | More Titrations / Discuss Midterm | Exp. 8 | Exp. 6 / Calculations |
| 12 M | 9/30 | The pH of High-Purity Water | pH meters | |
| 13 W | 10/02 | Acids, Bases, Buffers, Titrations | Exp. 7 | |
| 14 M | 10/07 | Midterm Review / Exp. 7 Report Format | ----- | Exp. 8 |
| 15 W | 10/09 | Midterm Exam | ----- | |
| 16 M | 10/14 | Electrochemistry | ----- | Exp. 7 |
| 17 W | 10/16 | Potentiometric Titrations | Exp. 16 | |
| 18 M | 10/21 | NoteBook Assignment (NBA) Part I | Analytical Sampling | |
| 19 W | 10/23 | EDTA Titrations | Exp. 12 | |
| 20 M | 10/28 | EDTA Titrations | Exp. 12 | NB Pages |
| 21 W | 10/30 | Spectrophotometry and Calibrations | Exp. 12 | Exp. 16 |
| 22 M | 11/04 | Spectrophotometry and Calibrations | Exp. 20 | |
| 23 W | 11/06 | Spectrophotometry and Calibrations | Exp. 20 | |
| 24 M | 11/11 | Analytical Separations | ----- | Exp. 12 |
| 25 W | 11/13 | Analytical Separations | Exp. 27 | |
| 26 M | 11/18 | More Spectrophotometry | IA Lab Tour | Exp. 27 |
| 27 W | 11/20 | How Much Caffeine is in Mountain Dew? | ----- | Exp. 20 |
| | 11/25-30 | Fall Break | ----- | |
| 28 M | 12/02 | More Calibrations/ NBA-II/Course Review | Exp. 23 / Drawers | |
| 29 W | 12/04 | Final Exam-Review | Exp. 23 / Drawers | |
| Final Exam | 12/11/2019 | Wednesday 11.00 AM – 1.45 PM | SLC 2.302 | Exp. 23 |

Course Policies

| | Harris Exp # | Title | 9th Edition PDF Page # |
|----------------------|---|---|------------------------|
| Experiments | 6. | Preparing Standard Acids and Bases | 34 |
| | 8. | Analysis of a Mixture of Carbonate and Bicarbonate | 40 |
| | 7. | Using a pH Electrode for an Acid-Base Titration | 37 |
| | 16. | Potentiometric Halide Titration with Ag ⁺ | 71 |
| | 12. | EDTA Titration of Ca ²⁺ and Mg ²⁺ in Natural Waters | 58 |
| | 20. | Spectrophotometric Determination of Iron in Vitamin Tablets | 83 |
| | 27. | Properties of an Ion-Exchange Resin | 102 |
| | 23. | Spectrophotometric Analysis of a Mixture: Caffeine & Benzoic Acid | 90 |
| Safety | <p>IMPORTANT: In accordance with University and Chemistry Department safety rules, any time anyone (student, TA, instructor, or visitor) is in a lab, Z87-rated safety eyewear must be worn. The first violation in the semester will result in a warning and removal from the lab until the safety eyewear is in-place. The second violation in the semester will result in dismissal from that lab period with no extra time being allowed for make-up of the work scheduled for that lab period. Similar penalties will apply if any other safety rules are violated. In summary, all students are responsible for all information inside the undergraduate safety manual uploaded on eLearning.</p> <p>In addition, arms, legs and feet should be covered in lab. Short pants and skirts (which expose calves or thighs) are not allowed. Sleeveless shirts (including spaghetti strap shirts), or shirts that expose your midriff are also not allowed—however, a lab coat may be worn over these shirts during lab. Closed-toed shoes that fully cover your foot are also required. Hair longer than shoulder length must be put up in an appropriate manner to keep it out of harm's way. Lab coats will be obtained from the dispensing machines in the BE building. The instructor will provide you with an access card for dispensing lab coats. Lab coats are single use and will be put in the soiled bin upon completion of the lab.</p> | | |
| Lab Reports | <p>Each student will prepare their own Lab Report for all 8 experiments based on the guidelines described in the Handout "Writing a Laboratory Report". The Lab Report for Exp. 6 will be evaluated but it will not count towards your grade. The remaining 7 Lab Reports are each worth 7 pts. Your 6 best Lab Reports will be summed for your final Lab Report Score (42 pts. Total).</p> <p>Please refer to the Class Schedule/Calendar for all Lab Report Due Dates. Lab Reports are due at the beginning of class. Late Lab Reports will be penalized at a deduction rate of 21% per week (3 % per day).</p> <p>If a student does not perform an Experiment, the student will receive zero (0) points for the corresponding Lab Report. Make-up of lab periods/experiments missed (for valid medical or emergency reasons) will be attempted based on the availability of the apparatus, BE 2.506, and the professor & TA.</p> | | |
| Lab NoteBooks | <p>Each student must bring his or her Lab NoteBook to UTD every Monday and Wednesday. Each student must keep his or her own neat and orderly Lab NoteBook using ink. Please put your name and a date on every NoteBook page you use. In addition, be sure to include data labels and units on all tables and graphs. Drawing chemical structures and balanced chemical reactions in your NoteBook is highly encouraged. Additional tips for keeping a professional NoteBook can be found on page 25 of your textbook. Your NoteBook must be signed and dated by your TA (or professor) at the end of any day you spend working in the lab.</p> | | |
| Lab Technique | <p>Each student will be evaluated with respect to their adherence to good safety practices, laboratory technical skills, and laboratory etiquette/professionalism. The evaluations will be made by your TA (with the professor) at the end of each Experiment (9 pts. Total). If one does not attend, one cannot earn Technique Points.</p> | | |

| Quizzes | The majority of Quizzes will be administered after the Midterm Exam and before the Cumulative Final Exam. There will not be make-up quizzes; a missed quiz equates to zero (0) points. There will also be one Take-Home Statistics Quiz in the first month. | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
|--------------------------------------|---|----|-------------|------------------------------|----|---------------|---|--------------------------------------|---|-------------------------|---|---------|---|--------------|----|------------------|-----------|-------|-----|----|------------|---|-------|---|-------|----|-------|----|-------|----|-------|----|-------|---|-------|---|-------|----|-------|----|-------|---|------------|----|-------|--|--|
| Midterm Exam | The Midterm Exam (Wednesday, October 09, 2019) will focus on Equilibrium, Acids and Bases, pH and pKa Calculations, Buffers, and Titration Curves. There will not be a make-up Midterm Exam; a missed Midterm Exam equates to zero (0) pts. Students must take the Midterm corresponding to the Section they are enrolled in. | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Final Exam | The Final Exam is <i>Cumulative</i> There will not be a make-up Final Exam; a missed Final Exam equates to (0) pts. Students must take the Final Exam corresponding to the Section they are enrolled in. Section-001 Final = 12/11/2019 Wednesday 11.00 AM – 1.45 PM SLC 2.302 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Grading (credit) Criteria | <p>Summary of Points:</p> <table style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th></th> <th style="text-align: right;"><u>Pts.</u></th> </tr> </thead> <tbody> <tr> <td>Lab Reports & NoteBook Pages</td> <td style="text-align: right;">42</td> </tr> <tr> <td>Lab Technique</td> <td style="text-align: right;">9</td> </tr> <tr> <td>Special NoteBook Assignment (Exp.12)</td> <td style="text-align: right;">2</td> </tr> <tr> <td>Buret Calibration Graph</td> <td style="text-align: right;">4</td> </tr> <tr> <td>Quizzes</td> <td style="text-align: right;">8</td> </tr> <tr> <td>Midterm Exam</td> <td style="text-align: right;">11</td> </tr> <tr> <td>Cumulative Final</td> <td style="text-align: right;"><u>24</u></td> </tr> <tr> <td>Total</td> <td style="text-align: right;">100</td> </tr> </tbody> </table> <p style="text-align: center;"><i>Quizzes, the Midterm, and The Final will be different for each Section.</i></p> <p>Your final letter grade for the course will be determined using a scale such as the one below where the class average is set at the “B-/C+” border (e.g., 79.5 points):</p> <table style="margin-left: auto; margin-right: auto;"> <tbody> <tr> <td>A+</td> <td>97 & above</td> <td>C</td> <td>73-76</td> </tr> <tr> <td>A</td> <td>93-96</td> <td>C-</td> <td>70-72</td> </tr> <tr> <td>A-</td> <td>90-92</td> <td>D+</td> <td>67-69</td> </tr> <tr> <td>B+</td> <td>87-89</td> <td>D</td> <td>63-66</td> </tr> <tr> <td>B</td> <td>83-86</td> <td>D-</td> <td>60-62</td> </tr> <tr> <td>B-</td> <td>80-82</td> <td>F</td> <td>59 & below</td> </tr> <tr> <td>C+</td> <td>77-79</td> <td></td> <td></td> </tr> </tbody> </table> <p style="text-align: center;"><i>Sections -001 & -002 & -003 & -004 are unique courses and are not graded together.</i></p> | | <u>Pts.</u> | Lab Reports & NoteBook Pages | 42 | Lab Technique | 9 | Special NoteBook Assignment (Exp.12) | 2 | Buret Calibration Graph | 4 | Quizzes | 8 | Midterm Exam | 11 | Cumulative Final | <u>24</u> | Total | 100 | A+ | 97 & above | C | 73-76 | A | 93-96 | C- | 70-72 | A- | 90-92 | D+ | 67-69 | B+ | 87-89 | D | 63-66 | B | 83-86 | D- | 60-62 | B- | 80-82 | F | 59 & below | C+ | 77-79 | | |
| | <u>Pts.</u> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Lab Reports & NoteBook Pages | 42 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Lab Technique | 9 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Special NoteBook Assignment (Exp.12) | 2 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Buret Calibration Graph | 4 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Quizzes | 8 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Midterm Exam | 11 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Cumulative Final | <u>24</u> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| Total | 100 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| A+ | 97 & above | C | 73-76 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| A | 93-96 | C- | 70-72 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
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| B+ | 87-89 | D | 63-66 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| B | 83-86 | D- | 60-62 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| B- | 80-82 | F | 59 & below | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| C+ | 77-79 | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |

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| Make-up Exams | <i>vide supra</i> |
| Extra Credit | None |
| Late Work | <i>No assignments will be accepted after the conclusion of “Final Exams Week”.</i> |
| Special Assignments | Students are financially responsible for items checked-out of the stockroom. |
| Class Attendance and Citizenship | <i>If a student is enrolled in Section-001, that student cannot attend Section-002 or -003 or -004 meetings (and vice versa). In addition, it is typical for CHEM-2401 activities to utilize the entire 225 minutes of class time such that students cannot simultaneously enroll in other classes whose meeting days and times conflict with those of the CHEM 2401 section they are enrolled in.</i> |
| Technical Support | If you experience any problems with your UTD account you may send an email to: assist@utdallas.edu or call the UTD Computer Helpdesk at 972-883-2911. |
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| Other Official UT Dallas Policies and Procedures | <i>The information contained in the following link constitutes the University's policies and procedures segment of the course syllabus.</i> http://go.utdallas.edu/syllabus-policies |
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These descriptions and timelines are subject to change at the discretion of the Professor.