

Course CHEM 5331 Advanced Organic Chemistry I

Professor Michael Biewer **Term** Fall 2016

Meetings T/R 4:00-5:15 PM, SLC 2.304

Professor's Contact Information

Office Phone 972-883-2811 Office Location BE 3.326

Email Address biewerm@utdallas.edu

Office Hours T/R 1-2 General Course Information

| General Course Information | | | | |
|------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|--|--|
| Pre-requisites, Co- requisites, & other restrictions | Undergraduate organic chemistry | | | |
| Course Description | Chemistry 5331 is intended to explain why organic molecules react as they do. The course is roughly divided into two broad topics: bonding (basis for bonds, where electrons are located, strength of bonds, etc.) and physical organic tools (linear free energy relationships, kinetics, isotope effects, etc.). By the end of the semester it is expected that the student can explain relative stability of molecules and be able to test how a reaction mechanism proceeds. There is no assigned text for the class. The course material may be downloaded from the webpage for this course [www.utdallas.edu/~biewerm, under teaching section click on CHEM 5331 link for access to material]. If anyone desires additional material to read a useful textbook option is Francis A. Carey and Richard J. Sundberg, "Advanced Organic Chemistry, part A", (recommend either 4 th or 5 th edition). The examinations will cover the material presented in class and the weekly packets handed out in class. I will periodically give packets containing suitable problems. These packets will not count against your grade. They are intended to give you a guide to what I expect and will be similar to problems on the tests. Therefore I strongly encourage you to work through the packets to point out difficult areas before the | | | |
| | tests. Two examinations will be given in this course along with a final examination. This course is a graduate course in chemistry and therefore a general knowledge of undergraduate organic chemistry is expected. Most topics traditionally covered in undergraduate organic chemistry will be only briefly reviewed. Students are encouraged to review selected topics in undergraduate books to understand the lectures for this course. | | | |
| Learning Outcomes | Upon completing this course, students will: • Be able to construct and apply both valence bond and molecular orbital | | | |
| Required Texts & Materials | None | | | |
| Suggested Texts, Readings, & Materials | Francis A. Carey and Richard J. Sundberg, "Advanced Organic Chemistry, part | | | |

Assignments & Academic Calendar

[Topics, Reading Assignments, Due Dates, Exam Dates]

| Date | | Topic | Group #* | Pages* |
|--------|---------------|------------------------------------------------|----------|---------|
| AUG 23 | 25 | Valence Bond Theory | 1 | 1-42 |
| 30 | SEP 1 | Valence Bond Theory, X-ray diffraction | 1,2 | 43-66 |
| 6 | 8 | Molecular Orbital Theory | 3 | 67-96 |
| 13 | 15 | Hückel Theory | 4,5 | 97-129 |
| 20 | | Applications | 5 | 130-139 |
| | SEP 22 | EXAM #1 | | |
| SEP 27 | 29 | Stereochemistry | 6 | 140-170 |
| OCT 4 | 6 | Thermochemistry/Conformation Analysis | 7,8 | 171-221 |
| 11 | 13 | Linear Free Energy Relationships | 9 | 222-238 |
| 18 | 20 | Kinetics | 10 | 239-260 |
| 25 | 27 | Kinetic Isotope Effects | 11 | 261-287 |
| NOV 1 | | Kinetic Isotope Effects | 11 | |
| | NOV 3 | EXAM #2 | | |
| NOV 8 | 10 | Nucleophilic Reactions | 12 | 288-311 |
| 15 | 17 | Nucleophilic Reactions/Cationic Rearrangements | 12,13 | 312-337 |
| 29 | DEC 1 | Pericyclic Reactions | 14 | 338-358 |
| DEC 6 | | Review | | |
| | | FINAL (Time and Date TBD) | | |

^{*} Group # and pages refer to the uploaded presentation number and the associated pages in those uploaded presentations for each topic located on the course website

Course Policies

| Course I officies | | |
|------------------------------|----------------------------------------------------------------------------------------------------------------------------------|--|
| Cuading (anadit) | Grades are based on the exams and final grades | |
| Grading (credit) Criteria | Tests $(2 \times 30\%) = 60\%$ | |
| | Final = 40% | |
| Make-up Exams | up Exams No make-up exams will be given | |
| | This creed was voted on by the UT Dallas student body in 2014. It is a standard that Comets | |
| Comet Creed | choose to live by and encourage others to do the same: | |
| Connet Creeu | | |
| | "As a Comet, I pledge honesty, integrity, and service in all that I do." | |
| | The information contained in the following link constitutes the University's policies and | |
| UT Dallas | UT Dallas procedures segment of the course syllabus. | |
| Syllabus Policies | | |
| and Procedures | Please go to http://go.utdallas.edu/syllabus-policies for these policies. | |
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The descriptions and timelines contained in this syllabus are subject to change at the discretion of the Professor.